

POROUS CARBON PELLETS FROM KOH TREATED SELF-ADHESIVE CARBON GRAINS FROM OIL PALM EMPTY FRUIT BUNCH: EFFECT OF KOH WEIGHT PERCENTAGES AND PELLETIZING COMPRESSION FORCE

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ABSTRACT

The pre-carbonized fibers of oil palm empty fruit bunch were ball milled to produce self-adhesive carbon grains (SACG). The SACG and SACG treated with KOH (1, 3 and 5% by weight) were compressed with various compression forces in order to produce green pellets with a dimension of 27 mm in diameter and 3 mm in thickness. Carbon pellets were produced from carbonization of green pellets (GP) up to 700°C in a nitrogen atmosphere using a multi-steps heating profile. The carbon pellets were immersed in HCl (0.5 M) for 48 h and then neutralized with distilled water. It was found that the BET surface area of carbon pellets of 0% KOH unchanged with compression force whereas carbon pellets from the GP of 5% KOH compressed with 6, 8 and 10 x 10³ kg forces have the BET surface area of 733, 399 and 361 m²/g respectively. The BET surface area of the unwashed carbon pellet of 0% KOH (2.3904 m²/g) was found to be 134 times smaller than that of the washed sample (321 m²/g). It was observed that as the percentage of KOH increased from 0 to 5% the BET surface area of the carbon pellets increased remarkably from 321 to 399 m²/g. Pore size analysis showed that the KOH treated carbons are mainly microporous. A SEM study of the KOH treated carbons showed increasing porous structures as the percentages of KOH increased and the compression forces decreased.

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REFERENCES

- [1] Bansal, R.C., Donnet, J.B., Stoeckli, F. (1988); Active carbon. Marcel Dekker, New York.
- [2] Hayashi, J. (1999); Carbon **37**, 524.
- [3] Teng, H.S., Chang, Y.J., & Heieh, C.T. (2001); Carbon **39**(13), 1981. [4] Park, S.J., & Jung, W.Y., (2002); Carbon; **40**, 2021.
- [5] Molina-Sabio, M., Rodriguez-Reinoso, F., Caturla, F., & Selles, M.J. (1996); Carbon **34**(4), 457.
- [6] Ahmadpour, A. & Do, D.D. (1997); Carbon **35**(12), 1723.
- [7] Smisek, M. & Cerny, S. (1970); Active carbon: manufacture, properties and applications. New York, Elsevier.
- [8] Sing, K.S.W., Everett, D.H., Haul, R.A.W., Moscou, L., pierotti, R.A., Rouquerol, J., & Siemieniowska, T. (1985); Pure Appl. Chem. **57**, 603.
- [9] Palm Oil Registration and Licencing Authority [PORLA] (1997); Review of the

- Malaysian Palm Oil Industry 1996; Ministry of Primary Industry, Malaysia Pg. 9. [10]
Mohamad, D. & Ramli, O. (1997); *Pertanika J. Sci. & Tech. (UPM)* **5**(1), 1.
- [11] Mohamad, D., Ramli, O. & Abdul Ghani, H. (1998); *J. Mat. Sci. Letters (U.K. /U.S.A.)* **17**, 2059.
- [12] Mohamad, D., Sarani Z., Ramli, O. & Astimar A. A. (2000); *Jpn. J. Appl. Phys.* **39**, 1236.
- [13] Mohamad, D., Ramli, O., Sarani Z., Izan Roshawaty M., Marina T., Norhayati, A. (2002); *J. Mat. Sci. Letters (U.K./U.S.A.)*. **37**, 1.
- [14] Mohamad, D., Sarani Z., Mohamad, H., Astimar A. A., Ridzuan, R., Anis M., Mohd. Nor M. & Mohd. Hamami, S. (1999); *J. Mat. Sci. Letters (U.K. /U.S.A.)* **18**, 249.
- [15] Byrne, C.E., & Nagle, D.C., (1997); *Carbon* **35**, 259. [16]
Mohamad, D. (1994); *J. Phys. D: Appl. Phys.* **27**, 1060.
- [17] Gregg, S.J. & Sing, K.S.W., (1982); *Adsorption, surface area and porosity* Academic Press.
- [18] Barret, E.P., Joyner, P.B. & Halenda, P. (1951); *J. Am. Chem. Soc.* **73**: 373. [19]
Ahmadpour, A, & Do, D.D., (1996); *Carbon* **34**(4), 471.
- [20] Hu, Z. & Srinivasan, M.P. (1999); *Microporous Mesoporous Mater* **27**,11. [21]
Teng, H. and Wang, S.C. (2000); *Carbon* **38**, 817.